

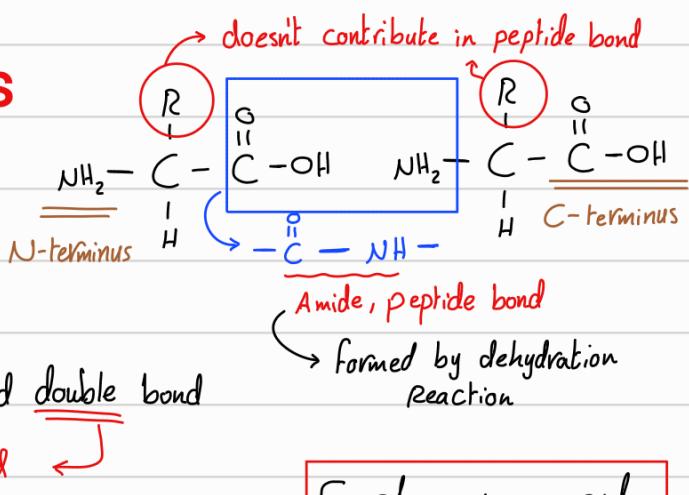
# Peptides

★ Peptide  $\rightsquigarrow$  more than one amino acids linked by peptide bond

$\rightarrow$  zigzag structure

$\rightarrow$  Resonance: shifting between single and double bond

No rotation  $\leftarrow$  Rigid  $\rightarrow$



★ Backbone  $\rightsquigarrow$  H,  $\alpha$ -C, NH<sub>2</sub>, Carbonyl

★ Side chain  $\rightsquigarrow$  R

★ All amino acids can form H-bonds except proline  $\rightsquigarrow$  because it is a secondary amine

Each amino acid is about 110 dalton

## Examples on peptides

### 1) Carnosine

$\rightarrow$   $\beta$ -alanine + L-Histidine

$\rightarrow$  Amino group on  $\beta$ -Carbon

$\rightarrow$  Protection against ROS and Muscle contraction

### 4) Oxytocin, Vasopressin

$\rightarrow$  Cyclic, 9 amino acids

$\rightarrow$  by disulfide bond

between 1-6 A.A

$\rightarrow$  have amide group on C-terminus to increase its half life and stability

### 2) Glutathione

$\rightarrow$   $\gamma$ -glutamate + Cystein + Glycine

$\rightarrow$  Amino group on  $\gamma$ -Carbon

$\rightarrow$  Scavenging oxidizing agents by forming disulfide bonds with Cys of other molecule

### ★ Oxytocin

$\rightarrow$  3  $\rightsquigarrow$  Ile, 8  $\rightsquigarrow$  Leu

$\rightarrow$  contraction of uterine

### ★ Vasopressin (ADH)

$\rightarrow$  3  $\rightsquigarrow$  Phe, 8  $\rightsquigarrow$  Arg

$\rightarrow$  Vasoconstriction, Water Retention,  $\uparrow$  BP

### 3) Enkephalins

$\rightarrow$  Tyr + Gly + Gly + Phe + Met

$\rightarrow$  Analgesics (pain killer)

$\rightarrow$  Similar to opiate (morphine) due to aromatic (Tyr, Phe) residues

### 5) Aspartame

$\rightarrow$  has a methyl Ester (methanol) on C-terminus

$\rightarrow$  L-Aspartate + L-Phe

$\rightarrow$  artificial sweetner

$\rightarrow$  D-isomer  $\rightsquigarrow$  Bitter

Must be avoided for PKU patient

Phe hydroxylase defect

phenylpyruvate accumulate

Mental retardation

Alatame  $\rightsquigarrow$  Aspartate + alanine

Poly peptide  $\rightsquigarrow$  long chain of amino acids, without 3D structure

Protein  $\rightsquigarrow$  1 or more polypeptide chains with a functional 3D structure

Native Conformation

## Primary Structure

- ★ Sequence of amino acids (polypeptide chain)
- ★ determine other levels
- ★ Isoforms are proteins have similar function with a slight difference on primary structure
- ★ Changes in the primary structure causes malfunctioning Proteins

- 1) Sickle cell anemia (HbS)
  - $\beta$ -chain, 6<sup>th</sup> A.A changed
  - Glu → Val
  - defect in RBCs
- 2) Cystic fibrosis

## Secondary structure

- ★ arrangement and rotation of backbone, stabilized by H-bonds

Regular  Non-Regular

- 1)  $\alpha$ -helix
  - helical rod with 3.6 AA per turn and 5.4  $\text{A}^\circ$  (pitch)

- ★ R-group outward

- ★ linear H-bonds

- ★ X Glycine

- ★ X proline

- ★ X Close similar charges

- ★ X  $\beta$ -branches (Val, Ile, Thr)

- 2)  $\beta$ -sheets

Loops

long

No conserved structure

$\beta$ -turns

Short, compacted

Glycine + Proline are common

- ★ 2 or more straight chains H-bonded side by side

- ★ Can be parallel or anti-parallel

- ★ X proline

- ★ Aromatic +  $\beta$ -branches are common

## Tertiary structure

Overall polypeptide 3D structure

Interactions between R groups

- ★ Shape determining interactions (Non Covalent)

- 1) H-bonds
- 2) Electrostatic
- 3) Van Der Waals
- 4) Hydrophobic

The most important

## Stabilizing factors

- 1) disulfide bond between thiol groups of 2 Cys forming Cystine

- 2) Metals → Iron in myoglobin (covalent)  
Zinc in Carbonic anhydrase (salt bridges)

## Quaternary structure

Overall structure of protein with

many polypeptide chains (subunits)

- ★ Subunits are linked by disulfide and non-covalent bonds
- ★ Immunoglobulin, Hemoglobin have Quaternary structure

Super-Secondary structure: a region of the polypeptide with many secondary structures

### 1) Motif

Repetitive secondary structures

- structural modules
- not related to function

- 3) Fold  $\rightarrow$  many domains with a specific function

### 2) Domain

a large region which fold and function independently on the rest of the protein

Fold > Domain > Motif

Denaturation: disruption of the native conformation by breaking non-covalent and disulfide bonds

The protein loses its properties and function, become insoluble (form aggregates)

1) Heat  $\rightarrow$  break Van Der Waals

Denaturing agents: 2) Extreme pH  $\rightarrow$  break electrostatic and H-bonds

3) Urea and guanidine hydrochloride  $\rightarrow$  break hydrophobic and H-bonds

4) Detergents  $\rightarrow$  Triton X-100 (uncharged)  $\rightarrow$  break hydrophobic

$\rightarrow$  SDS (anionic)  $\rightarrow$  break hydrophobic & H-bonds

all A.A Trans  $>$  Cis

except proline

5) Reducing agent ( $\beta$ -mercaptoethanol, dithiothreitol)  $\rightarrow$  break disulfide

Renaturation  $\rightarrow$  Returning the native conformation

$\rightarrow$  Chaperon  $\rightarrow$  help misfolded proteins to refold, so prevent aggregation

$\rightarrow$  Cis-trans isomerase  $\rightarrow$  shifting between Trans and Cis conformations

$\rightarrow$  Protein disulfide isomerase  $\rightarrow$  break and reform disulfide bonds

★ Misfolded when accumulated  $\rightarrow$  form aggregates, leading to disorders:

1) Prion disease  $\rightarrow$  transmissible

misfolded prion disease

PrP<sup>C</sup> ( $\alpha$ -helix)  $\rightarrow$  PrP<sup>SC</sup> ( $\beta$ -sheets)

Examples: Creutzfeldt - Jacob disease (human)

Scrapie (sheep), Mad Cow disease

2) Alzheimer disease

$\rightarrow$  not transmissible

accumulation of tau and amyloid

protein due to misfolding

in the APP

Apoprotein  $\rightarrow$  protein contains only amino acids

Holo protein  $\rightarrow$  protein + Non protein (prosthetic group)

$\rightarrow$  Coenzymes (organic factors)

$\rightarrow$  Metals

$\rightarrow$  lipids, sugars, Heme, phosphate

